



ISOPA PRODUS PROGRAM STEWART

# “Walk the Talk”

Clorura de Metilen

## Clorura de Metilen:

- CLASIFICARE :

**Evidenta limitata a efectului carcinogenic:**

**Fraza risc:**

**Xn R40**

**Fraze de siguranta:**

23 Nu respirati gaz/fum/vapori/spray

24/25 Evitati contactul cu pielea si ochii

36/37 Purtati haine si manusi de protectie potrivite



## Clorura de Metilen: Proprietati Riscante

### ▪ PRINCIPALE PREOCUPARI PRACTICE:

→ Mari concentratii de vapori vor cauza pierderi de simturi (anestezie) si inconstienta (narcosia)

→ Iritarea pielii prin contact direct

### → ALTE PROPRIETATI DE AVERTIZRARE:

-> Miros dulce la un nivel destul de ridicat: avertizare pentru expuneri riscante.

-> Densitate gaz >>> aer: vaporii tind sa raman localizati sau sa se disperseze incet in zona unde respira muncitorii

-> Scara de inflamabilitate: 14% la 22% (in aer) : risc relativ scazut



## Masuri de protectie

- Expunerea la nivele ridicate de clorura de metilen este probabila daca clorura de metilen sau un produs care o contine, este folosit intr-o camera cu ventilatie necorespunzatoare.
- Cand e incalzit la descompunere, chiar si cu o tigara aprinsa, clorura de metilen poate genera fosgen si fluorura de carbon.

..... SO .....→

- Folositi PPE cand lucrati
- Verificati ca sistemul de extractie sa fie pornit
- Nu mancati, nu beti sau fumati la locul de munca
- Daca nu va simtiti bine, informati-va colegii si parasiti locul de munca

***Unde ati putea fi expus?***

# Unde ati putea fi expus?



- Zona de spumare
- Zona de debitare
- Zona de maturare
- Operatii de curatare folosind clorura de metilen ca solvent
- Scurgeri



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Pentane

## Hazardous Properties of Pentane

- Flash Point:  $-40^{\circ}\text{C}$  to  $-20^{\circ}\text{C}$   
(the lowest temperature at which liquid releases sufficient vapour for ignition)
- Auto ignition temperature ca.  $280^{\circ}\text{C}$   
(where the vapour-air mixture ignites on a hot surface.)
- Explosive vapour-air mixtures:  
Lower explosion limit:  $1,4 \text{ Vol}\% = 41 \text{ g/m}^3$   
Upper explosion limit  $7,8 \text{ Vol}\% = 240 \text{ g/m}^3$   
(Evaporation rate at  $20^{\circ}\text{C} - 30^{\circ}\text{C} > 2,4 \text{ kg/h per m}^2 \text{ surface}$ )  
Vapour has higher density than air!
- Easy build-up of electric charge.
- Highly flammable.



## Protective Measures

### Avoid explosive atmosphere (primary measure)

- No open handling, closed systems.
- Controlled ventilation.
- Generate inert atmosphere with Nitrogen.
- **Avoid sources of ignition (secondary measures)**
- Explosion protected machinery (encapsulation, no sparks, no hot surfaces)
- Avoid electrostatic sparks (earth machinery, no plastic containers)



## Incident / Alarm

- **In case of spillage or alarm from monitors**

Keep calm

- **Stop pentane dosage, avoid ignition sources**

Increase ventilation

Stop leakage

Cover spillage with absorbant

Self-contained breathing apparatus if ventilation is insufficient

Protective clothing

Extinguishing material: CO<sub>2</sub>, Foam, Powder; no water!



## Use of Pentane

Pentane is highly flammable and may build up explosive mixtures with air

- Avoid any ignition source
- Ensure that static electricity cannot build-up
- Monitor level of pentane in air; remember that pentane is heavier than air



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**Catalysts**

## Catalysts

- Catalysts can be corrosive, irritants, sensitisers and flammable
- Symptoms of exposure include chemical burn, swelling, itching, redness and hazy vision
- Use PPE when working with catalysts and polyol formulation components



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Methyl Isocyanates

## Monoisocyanates and Diisocyanates

- Monoisocyanates are used for various applications – but not for polyurethanes

For example, methyl isocyanate is used for pesticides and insecticides

- All polyurethanes are made with diisocyanates such as MDI or TDI

## Monoisocyanates and Diisocyanates

- Monoisocyanates are used for various applications – but not for polyurethanes.  
For example, they are used in pesticides.
- All polyurethanes are made from diisocyanates such as MDI or TDI.

*Methyl-isocyanate  
is not used in  
making polyurethanes*

## Disclaimer

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